

Essential Protein and Enzyme Demonstrations Worksheet

Demonstration #1 — pH and Protein Solubility

Data Table 1

| Chemical Added to Casein | Observations | Approximate pH |
|-----------------------------------|--------------|----------------|
| 250 mL of 0.01 M Sodium Hydroxide | | |
| 10–15 mL Hydrochloric Acid | | |
| 15–20 mL Hydrochloric Acid | | |
| 20–30 mL of Sodium Hydroxide | | |

Demonstration #2 — Digestive Enzymes at Work

Data Table 2

| Beaker Contents | Observations |
|-----------------------------|--------------|
| Protein, water, and biuret | |
| Protein, pepsin, and biuret | |
| Starch, water, and iodine | |
| Starch, amylase, and iodine | |

Demonstration #3 — The Floating Catalyst

Data Table 3

| Beaker | A | B | C | D |
|--|---|---|---|---|
| Concentration of H ₂ O ₂ | | | | |
| Average Reaction Time | | | | |
| Average Rate (1/Time) | | | | |

Questions

Demonstration #1

1. Casein has both acidic side chains and basic side chains. At a high (basic) pH, ionization occurs in the acidic chains. At a very low (acidic) pH, protonation occurs in the basic chains. Do you think casein is most soluble with a net charge that is positive, negative, or around zero? Why?
2. A protein's isoelectric point is the pH at which the protein has a net charge of zero. Approximate the isoelectric point of casein.

Demonstration #2

3. Compare and contrast the observations of the biuret test results. Describe the evidence, if any, for the digestion of protein using pepsin.
4. The pepsin solution was prepared using 0.01 M hydrochloric acid in order to optimize the pepsin enzyme. Why was this necessary?
5. Compare and contrast the iodine test results for starch and starch/amylase. Explain the test results based on the activity of amylase.

Demonstration #3

6. What is the purpose of catalase in the human body?
7. Create a graph comparing the Average Reaction Rate to the Concentration of H_2O_2 (%).
8. Examine the graph from Question 7. How does the reaction rate change at high versus low concentrations of hydrogen peroxide?