

# Series and Parallel Circuits Worksheet

## Series Circuit Observations

*One Lightbulb*

*Two Lightbulbs*

*Three Lightbulbs*

*Open Circuit*

## Parallel Circuit Observations

*Two Lightbulbs*

*Three Lightbulbs*

*Open Circuit*

*Two Lightbulbs and One Short Circuit*

# Series and Parallel Circuits Worksheet, continued

## Post-Lab Questions

1. Which circuit design produced the brightest lightbulbs? Relate this to the amount of current flowing through each lightbulb.
2. What happened when all three lightbulbs were connected in series? Why did this occur?
3. Is there more resistance in the series circuit or the parallel circuit? How can you tell?
4. What is one advantage of a series circuit? What is one disadvantage?
5. What is one advantage of a parallel circuit? What is one disadvantage?
6. What happened with the two parallel-connected lightbulbs and the short circuit?
7. Is it better to have a string of lights, such as Christmas-tree lights, connected in series, or parallel? Explain.

# Pith Ball Electroscope Worksheet

## Observations

Charge by Induction

Charge by Conduction

## Post-Lab Questions

1. What did the pith balls do when the positive and negative charges were brought near them? Did the pith balls respond differently to the positive and negative charges?
2. What did charging the electroscope by conduction do? (Optional) Draw a picture showing the charged electroscope.
3. After the electroscope was charged by conduction, what charge (positive or negative) did the electroscope carry? How do you know?
4. Explain why the pith balls immediately “fly away” from the charged rod after making contact.

# Measuring Cell Potentials Worksheet

## Data Table

Part A. Cell Potentials versus Zinc as the Reference Electrode				
Metal	Positive Electrode (Cathode)	Negative Electrode (Anode)	Measured Cell Potential $E_{\text{cell}}^{\circ}$ (V)	Calculated Value $E_{\text{red}}^{\circ}$ (V)
Copper				
Magnesium				

Part B. Predicted and Measured Cell Potential			
Cathode/Anode	Calculation (Equation 1)	Predicted Cell Potential	Measured Cell Potential

## Post-Lab Questions

- Which metal was most easily oxidized (it always appeared as the anode)? Which metal ions were most easily reduced (the corresponding metal always appeared as the cathode)?
- Rank the three metals tested (including zinc) from most positive to most negative standard reduction potential. Write a general statement describing the relationship between the standard reduction potential of a metal and metal activity.
- Look up the literature values of the standard reduction potentials for Cu and Mg, and calculate the percent error for each.  
*Hint:* Note the symbol for “absolute value.”

$$\text{Percent error} = \frac{|\text{Experimental value} - \text{Literature value}|}{\text{Literature value}} \times 100\%$$

# Resistance Worksheet

## Observations

“Simple Circuit” lightbulb brightness

Thick wire lightbulb brightness

Thin wire lightbulb brightness

Thin wire (1 m) lightbulb brightness

## Post-Lab Questions

1. How does the brightness of the lightbulb relate to the resistance in the circuit?
2. How did the length of the short wire affect the brightness of the lightbulb? What does this mean in terms of the resistance in the circuit?
3. How did the thickness of the wire affect the brightness of the lightbulb? What does this mean in terms of the resistance in the circuit?
4. What are the two physical dimensions that affect the resistance in a wire? How do these physical dimensions affect the resistance in a wire?