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## Mineral Rummy Worksheet

## **Post-Activity Questions**

- 1. A pearl is considered a gemstone and its composition includes calcium carbonate. Explain whether or not a pearl is a mineral.
- 2. Both hematite and magnetite are composed of the same two elements, iron and oxygen. Why are they considered different minerals?
- 3. Fill in the chart below by writing the chemical name of each mineral and the group to which it belongs.

| Mineral Name | Chemical Formula                    | Chemical Name | Mineral Group |
|--------------|-------------------------------------|---------------|---------------|
| Azurite      | $Cu_3(CO_3)_2(OH)_2$                |               |               |
| Barite       | BaSO <sub>4</sub>                   |               |               |
| Beryl        | $\mathrm{Be_{3}Al_{2}Si_{6}O_{18}}$ |               |               |
| Cinnabar     | HgS                                 |               |               |
| Iodargyrite  | AgI                                 |               |               |
| Periclase    | MgO                                 |               |               |
| Zircon       | ZrSiO <sub>4</sub>                  |               |               |

4. The chemical formula of malachite is Cu<sub>2</sub>(CO<sub>3</sub>)(OH)<sub>2</sub>. Given the respective charges of the carbonate and hydroxide anions in the *Background* section, determine the charge of the copper cation.

5. What might be the value of classifying minerals by their metal cations? What difficulties might one encounter using this classification system?