

Name

## Iodine Clock Reaction Demonstration Worksheet

**Results** Table

	1 (Control)	2	3	4	5	6
Contents of Beaker A						
Contents of Beaker B						
[KIO <sub>3</sub> ] after mixing solutions						
[Na <sub>2</sub> S <sub>2</sub> O <sub>5</sub> ] after mixing solutions						
Temperature						
Catalyst present?						
Reaction Time						

## **Discussion Questions**

- 1. Write the chemical equation for each of the steps in this reaction mechanism.
  - *a*. Sodium meta-bisulfite produces hydrogen sulfite ions in water
  - b. Potassium iodate decomposes
  - c. Iodate ions react with hydrogen sulfite ions
  - d. Iodide ions react with iodate ions in the presence of water

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2. How does the concentration of reactant molecules affect the reaction rate? Explain.

3. How does temperature affect the reaction rate? Explain.

4. What is a catalyst? What was the catalyst used in this demonstration?