

# Science Department Safety Training Notes

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#### **Discussion and Notes**

Keep a copy of these safety training notes and a signed attendance sheet to verify regular safety training. Regulatory inspectors will usually request proof of safety training. A copy of the sign-up sheet that we suggest using may be found at <a href="https://www.flinnsci.com/media/412875/signup.pdf">www.flinnsci.com/media/412875/signup.pdf</a>.

The important CSB safety message and the accompanying video may be viewed on the CSB website <a href="http://www.csb.gov/csb-releases-safety-message-on-preventing-accidents-in-high-school-chemistry-labs/">http://www.csb.gov/csb-releases-safety-message-on-preventing-accidents-in-high-school-chemistry-labs/</a>

OSHA revised the Hazard Communication Standard in 2012 to integrate objective Globally Harmonized System (GHS) standards for the classification and labeling of chemicals. The GHS hazard category rankings for flammable liquids are different than the NFPA system! To avoid confusion, always look for the descriptive GHS hazard statement on the label and SDS. The hazard statements are consistent, clear and concise!

All Flinn Scientific chemical labels and Safety Data Sheets are GHS-compliant. For free online access to all Flinn SDS, go to <a href="http://www.flinnsci.com/msds-search.aspx">http://www.flinnsci.com/msds-search.aspx</a>

# **Prevent Tragic Accidents with Flammable Liquids**

Girl in Flames Runs from School Lab Explosion! Lab Accident Sends Teacher and Students to Hospital! Teacher Charged After Experiment Goes Awry!

These real news headlines appeared within the past year as a result of accidents involving flammable liquids in middle school and high school science labs. Fires in school science labs are always tragic, all the more so because they can easily be prevented. In a Safety Alert issued after a recent school fire and explosion, the United States Chemical Safety Board (CSB) delivered a clear and unequivocal message—with better attention to good safety practices, and safety training, accidents like these can be avoided. The purpose of this safety note is to review essential safety practices for using flammable liquids.

### **General Safety Precautions**

Always keep flammable liquids away from heat, sparks, open flames and other sources of ignition. Use the smallest amounts possible in experiments and demonstrations, and always cap solvent bottles when not in use. Most flammable liquids are volatile organic compounds that are also slightly toxic by inhalation and are body tissue irritants. Mild headaches or dizziness may be symptoms of overexposure to organic vapors. Always work with flammable organic liquids in a hood or well-ventilated lab. Wear chemical splash goggles, chemical-resistant gloves, and a chemical-resistant apron.

### **Classifying Flammable Liquids**

OSHA classifies flammable liquids into four categories based on their flash points and boiling points. The flash point is the minimum temperature at which a liquid forms a vapor above its surface in sufficient concentration that it may be ignited with a spark or flame. GHS criteria and hazard statements for flammable liquids are shown in the following table, along with common examples.

GHS Hazard Statement	Flash Point	<b>Boiling Point</b>	Examples
Extremely flammable liquid and vapor	<23°C (73°F)	≤ 35°C (95oF)	Ethyl ether
Highly flammable liquid and vapor	<23°C	>35°C	Methyl alcohol, Ethyl alcohol, Acetone, Toluene, Hexane
Flammable liquid and vapor	≥ 23°C and ≤ 60°C	Not specified	Glacial acetic acid, n-Butyl alcohol
Combustible liquid	≥ 60°C and ≤ 93°C	Not specified	Kerosene

To understand these criteria in context, consider a liquid exposed to a spark. A liquid with a flash point less than 100°F (38°C) can be ignited by a spark at or near room temperature. Combustible liquids may require more than a spark to ignite them at room temperature, but are dangerous when heated. Holding a flame above a combustible liquid will quickly heat the vapors above the flash point and may start a fire.

The Safety Data Sheet (SDS) of a substance lists the GHS hazard category and the accompanying hazard statement, along with its flash point, NFPA code and hazard rating (if available), and the autoignition temperature. The autoignition temperature is the temperature at which the material will initiate a self-sustained fire even without an ignition source.

#### **Discussion and Notes**

Flinn Scientific maintains a comprehensive database of tested lab procedures for thousands of topics in the middle school and high school science curriculum. Please contact us by phone or email if you need an activity for a specific topic, principle or concept in your science class. We are delighted to share!

### Safety Checklist for Working with Flammable Liquids

- Work with flammable liquids in the hood or a well-ventilated lab equipped with a purge
  fan. The lab must also be furnished with an ABC fire extinguisher, fire blanket, safety eyewash, and safety shower.
- Never use flammable liquids around an ignition source. Organic liquid vapors are heavier than air and will quickly travel great distances along a lab bench or floor to an ignition source.
- For demonstrations involving highly flammable liquids and vapors, dispense the liquid in small amounts before beginning the demonstration.
- Cap the solvent bottle and remove it far away from the demonstration area.
- Never add fuel to a fire! Do NOT pour a flammable liquid onto a hot surface, flame or other ignition source, and NEVER add more liquid once a demonstration is underway.

#### **Always Follow a Written Procedure**

Comprehensive safety training goes beyond individual safety precautions for specific chemicals or events to build a culture of safety and truly establish "Safety First" in your school. Teachers who are certified in science lab safety know that before doing any lab activity they must identify all physical, chemical and biological hazards, evaluate their relative risk, and adopt safe WRITTEN procedures to protect against these hazards. You must have a written procedure before doing any lab activity!

## **Common Lab Fire Captured on Video**

We have reenacted a lab fire using methyl alcohol to reinforce the safety precautions described in this note and demonstrate that these accidents are totally preventable. Please click on the following link <a href="http://www.flinnsci.com/cla3">http://www.flinnsci.com/cla3</a> to view our dramatic Common School Laboratory Accidents video. It is only through education that we can ensure this type of accident will never occur again!

### **Safety Training Will Prevent Accidents**

Safety experts know that it's easy to prevent the LAST accident you may have heard about. How do you prevent the NEXT one? Make safety a leading indicator for educational success in your school by encouraging all science teachers to complete the Flinn Scientific Laboratory Safety Course. Please visit the Flinn safety site at <a href="http://labsafety.flinnsci.com/Home.aspx">http://labsafety.flinnsci.com/Home.aspx</a> to learn about this comprehensive safety education program, available online, absolutely free, for all science teachers!

#### Thank You for Your Support

Please continue to support our efforts to improve safety in school science labs by ordering your laboratory chemicals and science supplies from Flinn Scientific.

#### **Next Month's Topic**

**Take Control of Your Chemical Inventory** 

